

Using Free Energy Perturbation Calculations to Model the Mutation of LeuT_{Aa} and mDAT Residues that Bind TCAs

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Advisor: Dr. Jeffry Madura

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Free Energy Perturbation Calculations?

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Free Energy Perturbation (FEP) Calculations

- Can be used to predict manifold biological phenomena
 - pKa
 - change in binding energy due to mutagenesis
 - solvation energy
 - protein-ligand binding
- Usually the most accurate way to calculate any of the above

The Zwanzig Equation

$$\Delta G(A \rightarrow B) = G_B - G_A = -k_B T \ln \left\langle \exp \left(-\frac{E_B - E_A}{k_B T} \right) \right\rangle_A$$

Zwanzig, R.W. **1954**. *J. Chem. Phys.* 22; 8: 1420-1426.

Implementation in NAMD 2.7b1

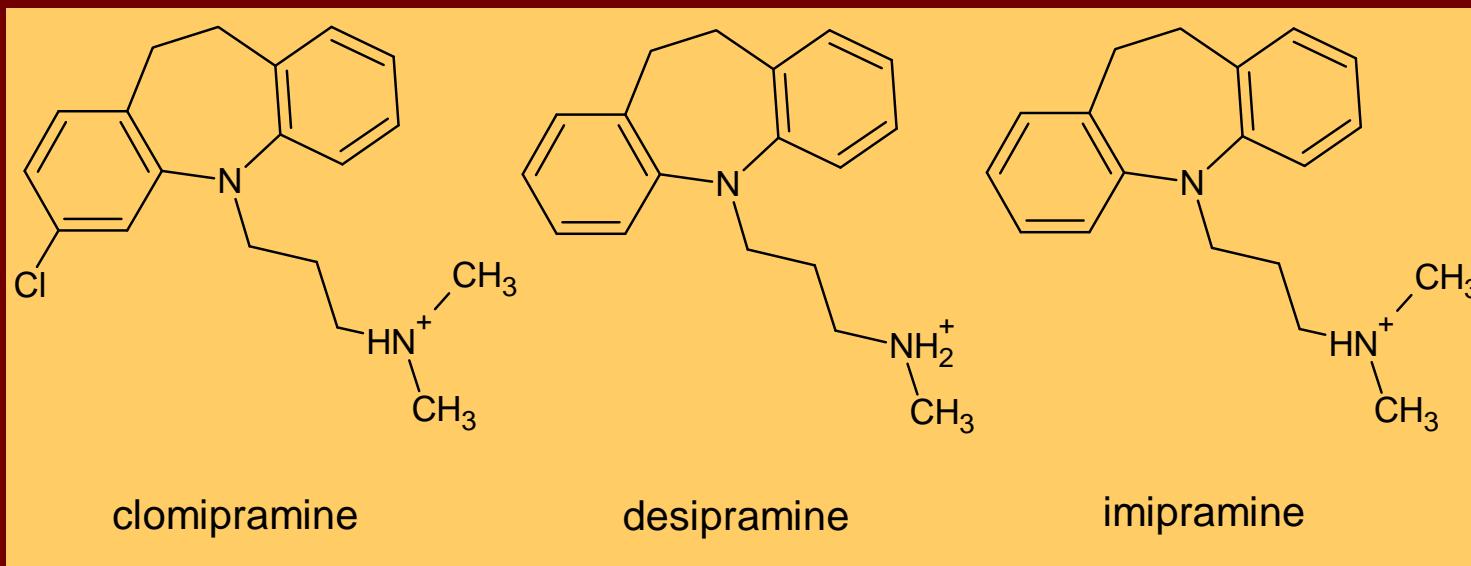
$$\Delta G(i \rightarrow i + \Delta i) = -k_B T \sum_{i=1}^N \ln \left\langle \exp \left(-\frac{H_B(x, p_x; \lambda_{i+\Delta i}) - H_A(x, p_x; \lambda_i)}{k_B T} \right) \right\rangle_i$$

Parameterizing a Novel Molecule

- Determines how a force field (e.g., CHARMM) will treat the molecule
 - Atomic Partial Charges
 - Bond lengths and angles
 - Spring Constants for bonds and angles
 - $K_b(b-b_0)^2$
 - Energy function for rotation around dihedral bonds
- Atomic Partial Charges were assigned via the OPLS-AA force field
- Remaining parameters were set via (educated?) guesses

LeuT and TCA Binding

- The crystal structure of LeuT_{Aa} was reported in 2005.
- LeuT_{Aa} was also crystallized bound to several tricyclic antidepressants (TCAs)



Yamashita, A. et al. **2005**. *Nature*. 437: 215-223.

Zhou, Z. et al. **2007**. *Science*. 317: 1390-1393.

Singh, S.K. et al. **2007**. *Nature*. 448: 952-956.

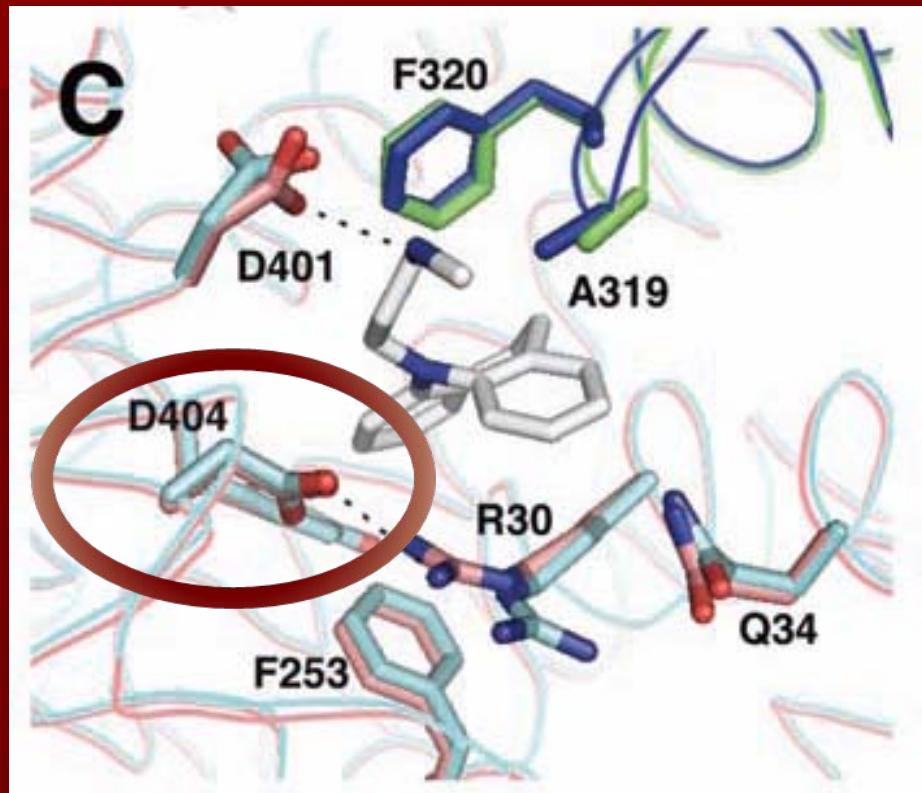
Varying Steps/Window

Mutated Residue: LeuT Asp 404 → Ala

Steps/Window	ΔG_{D404A} (kcal/mol)
100,000	114.4
200,000	115.2
400,000	119.7

Calculations performed in vacuum

The Mutation



Aspartate 404 mutated to alanine with
clomipramine bound

Modeling Ligand Binding with FEP Calculations



Modeling Ligand Binding with FEP Calculations



$$\therefore \Delta\Delta G_{bind} = \Delta G_2 - \Delta G_1$$

Initial TCA Binding Results

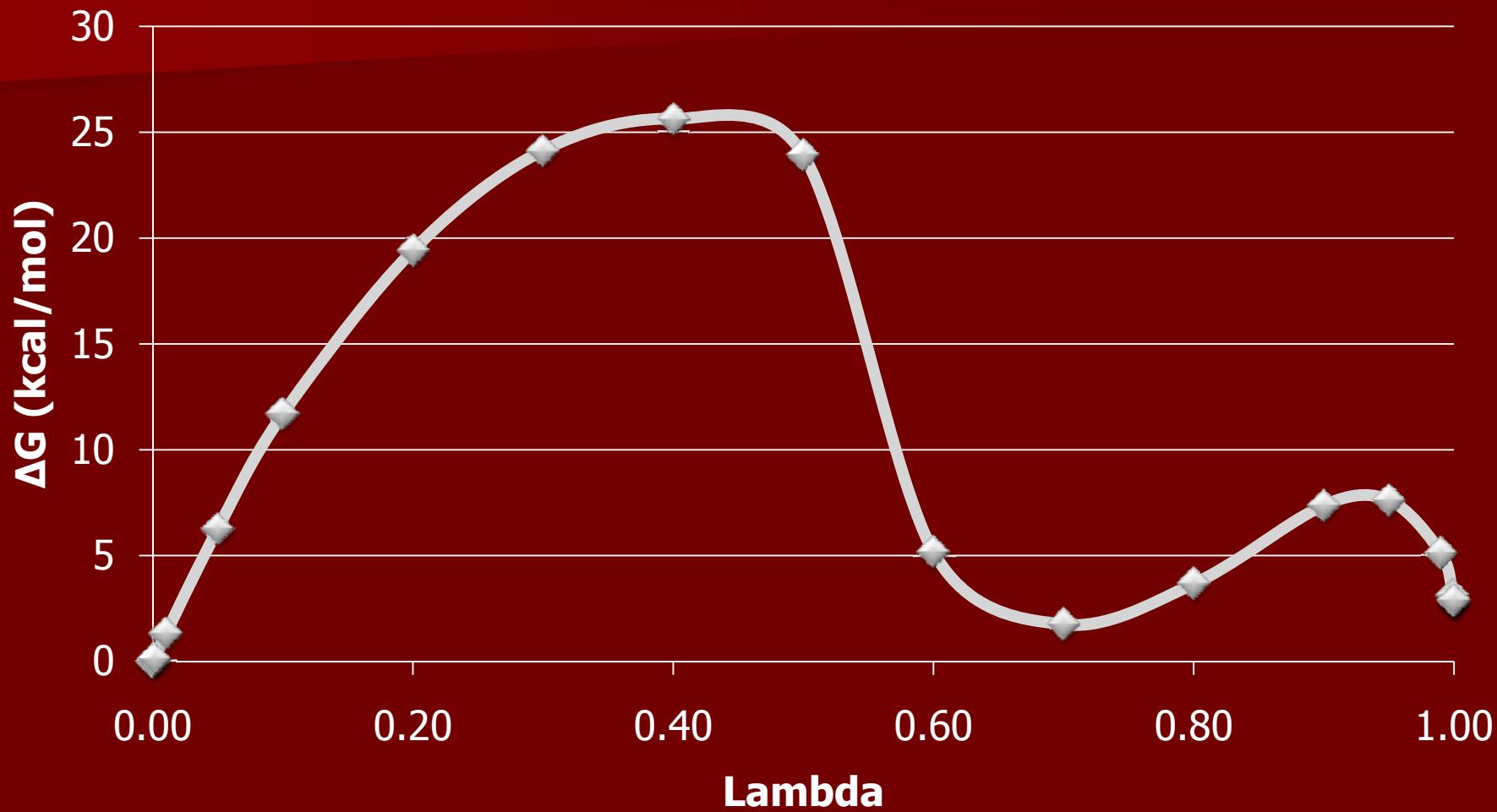
FEP Calculations of Aqueous $\Delta\Delta G_{\text{binding}}$ Results

TCA (Complex)	$\Delta G_{\text{TCA} \rightarrow \sim}$ (kcal/mol)	$\Delta G_{\text{LeuT:TCA} \rightarrow \text{LeuT:}\sim}$ (kcal/mol)	ΔG_{bind} (kcal/mol)	ΔG_{exp} (kcal/mol)
clomipramine	2.9	15	-12	-4.9 ± 0.09
imipramine	35	12	23	-3.7 ± 0.08

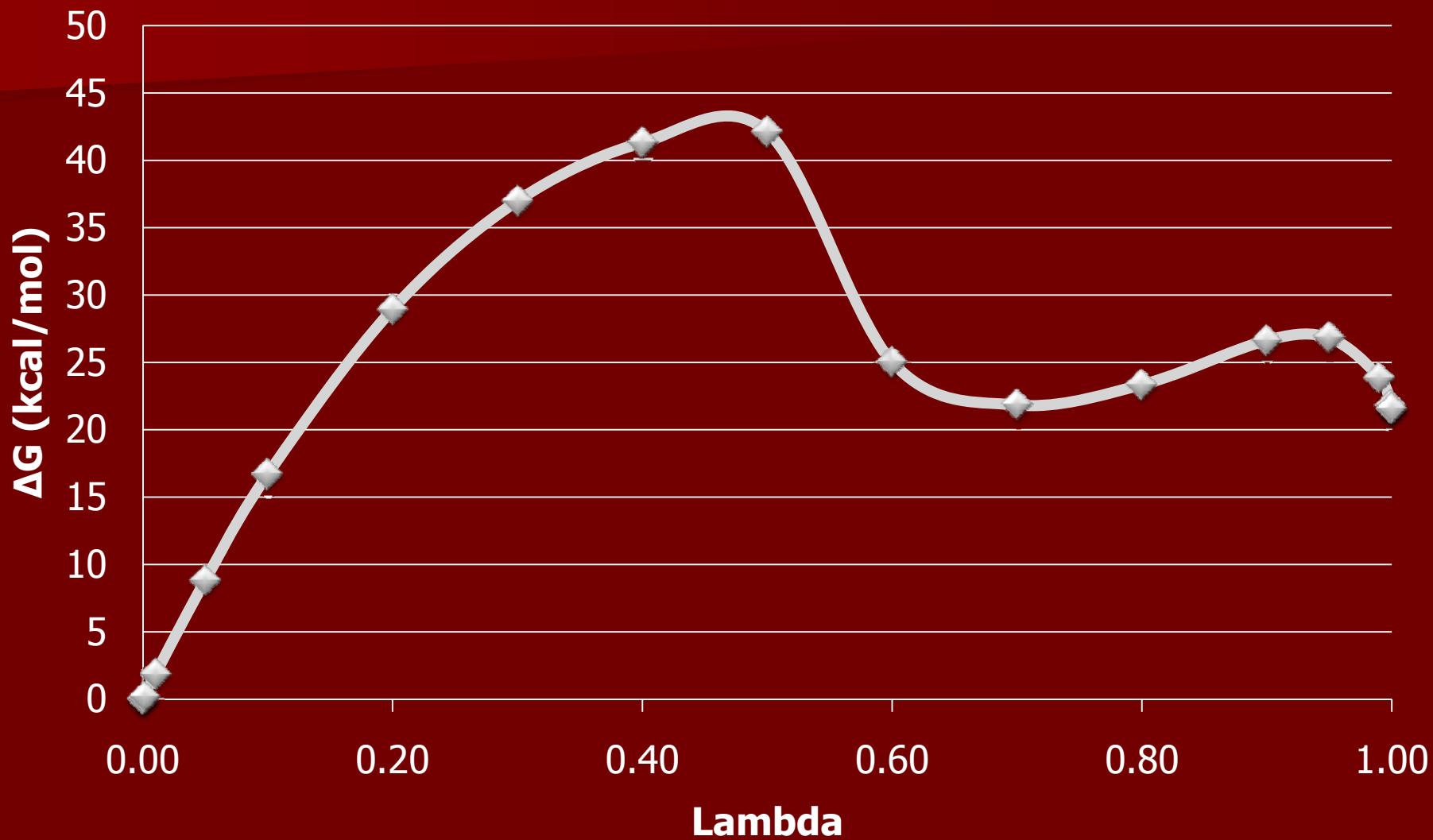
- $\Delta\Delta G_{\text{binding}} = RT\ln(\text{IC}_{50})$ for non- or un-competitive binding

Experimental values from Singh, S.K. et al. 2007.
Nature. 448: 952-956. (supplementary material)

Clomipramine $\rightarrow \sim$



Desipramine $\rightarrow \sim$

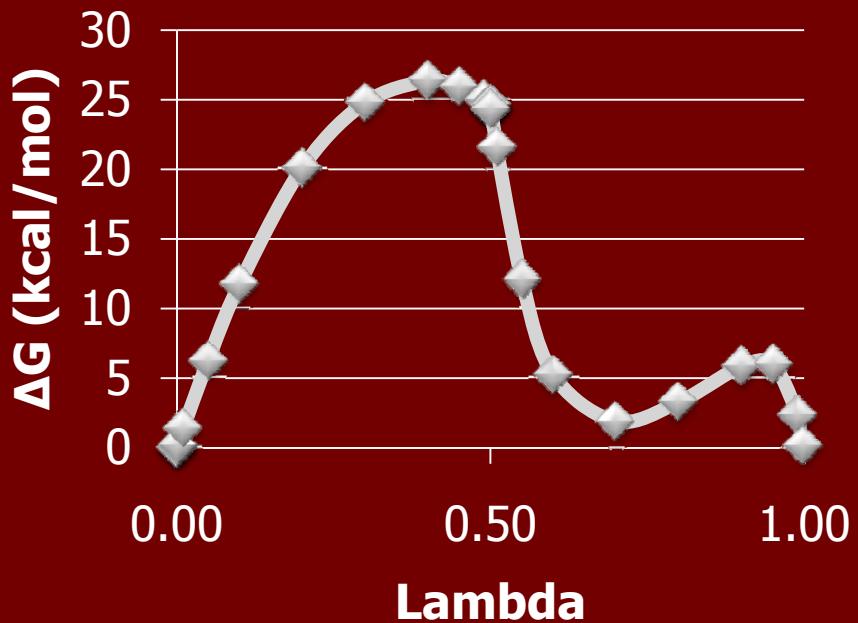


Potential Sources of Error

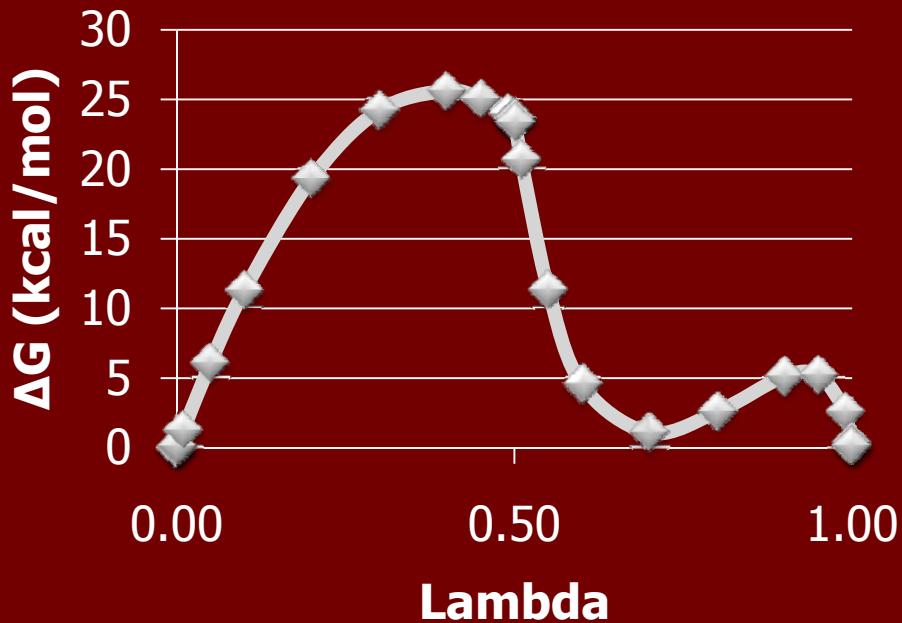
- Inadequate sampling near $\lambda = 0.5$
 - More trials run with better sampling (still waiting for results from supercomputer)
- OPLS charges incompatible with force field
 - Performed ab initio calculations to compare to OPLS charges
- FEP calculations incorrectly implemented
 - Ran several variations of FEP calculations using decane (with OPLS charges and guessed parameters)
 - Prepared a set of 14 calculations with CHARMM-parameterized molecules (will run when supercomputer wakes up)

TCA $\rightarrow \sim$ with Better Sampling

CXX $\rightarrow \sim$



IXX $\rightarrow \sim$

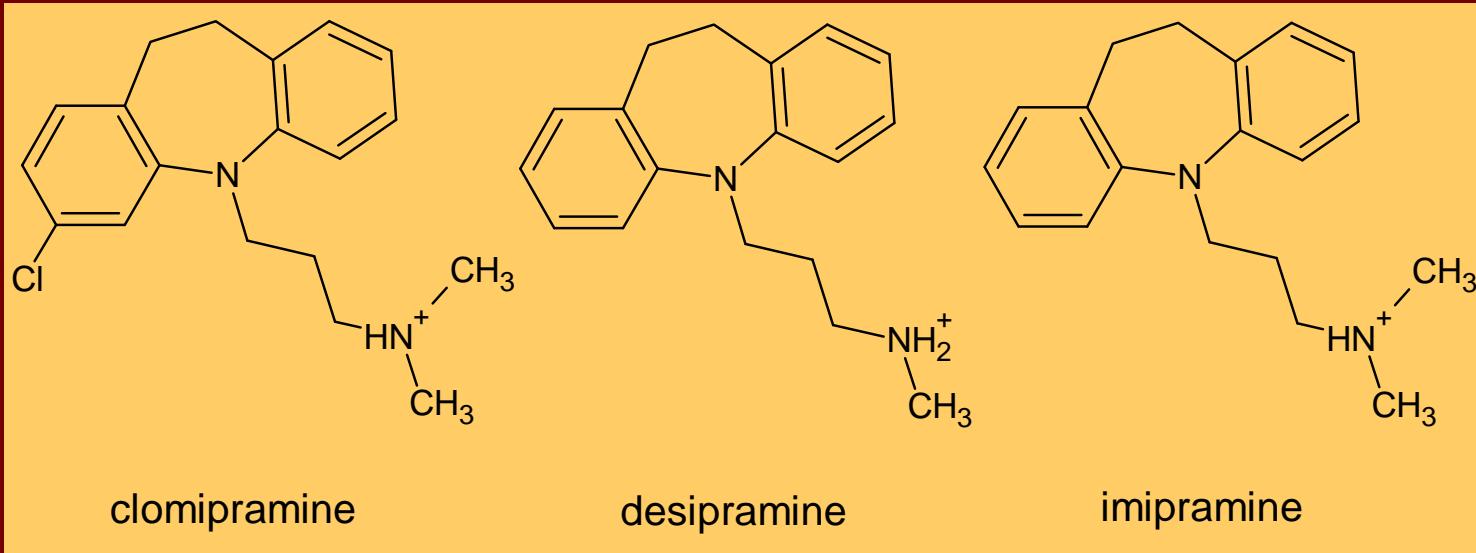


100K steps / window, 39 windows

Electrostatics decoupled $\lambda = 0 - 0.5$, van der Waals decoupled $\lambda = 0.5 - 1.0$

Comparison of TCA → ~ Calculations

	25 Windows (kcal/mol)	39 Windows (kcal/mol)
clomipramine	2.86	-0.0865
imipramine	35.0	0.184
desipramine	21.6	21.9

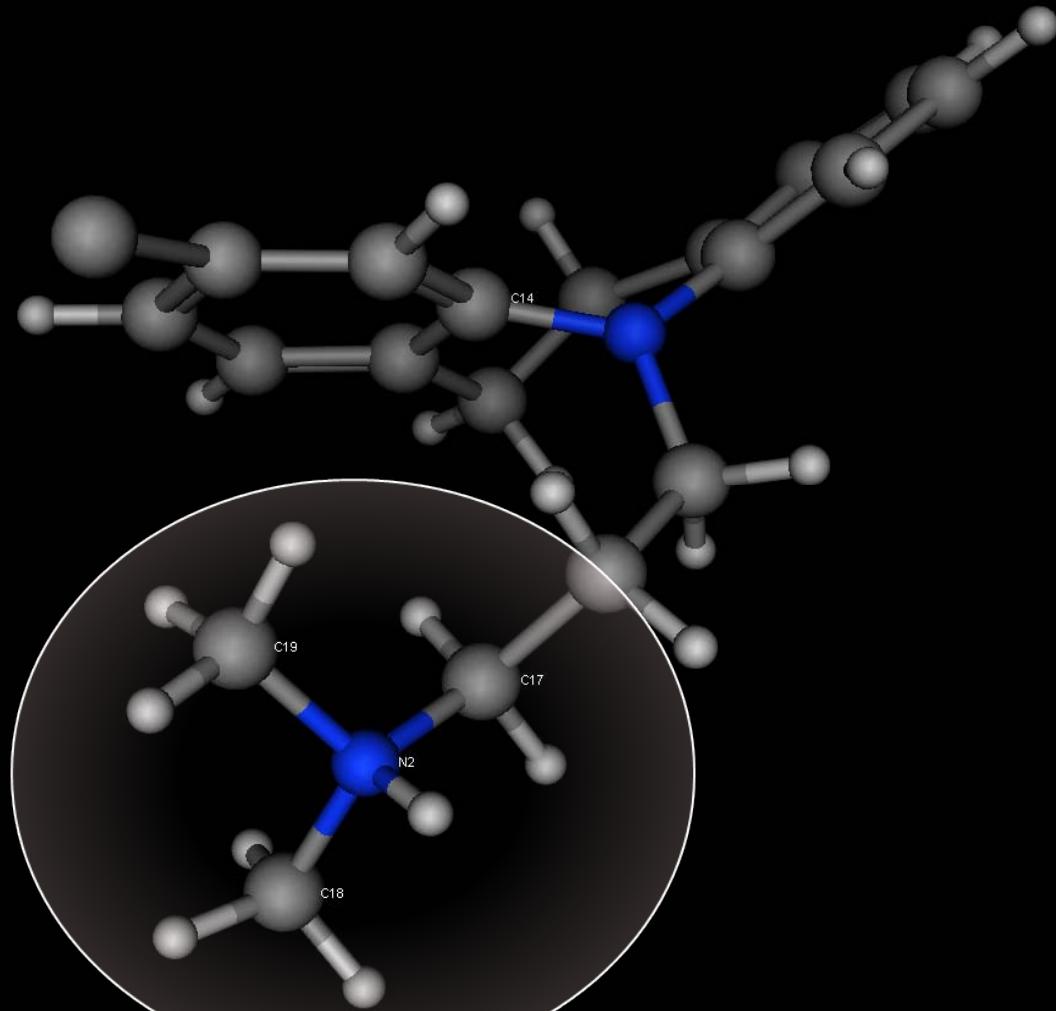


Alternative Charges

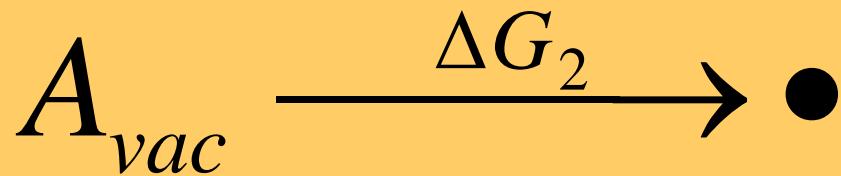
		Minimized Structure		XRD Structure	
	OPLS-AA	MSK	ChelpG	MSK	ChelpG
Total Charge	1.00	0.992	1.001	0.998	1.00
Abs. Dev / N	--	0.090	0.080	0.112	0.088
Most significant deviations					
Atom	Charges				
N2	-0.260	0.003	0.005	0.103	-0.058
C14	0.123	0.267	0.336	0.126	0.214
C17	0.190	0.319	0.124	-0.216	0.075
C18	0.130	-0.359	-0.217	-0.387	-0.262
C19	0.130	-0.327	-0.152	-0.438	-0.194

Calculated in Gaussian 03 at the HF/6-31G(d) level of theory; the minimized structure contained no imaginary frequencies.

Most Significant Deviations



Determining $\Delta\Delta G_{solv}$



$$\Delta\Delta G_{solv} = \Delta G_2 - \Delta G_1$$

Results for FEP Calculations on Decane

		Aqueous			
		Decouple	ON	OFF	
Fixed?		Yes	No	Yes	No
Boundary	Yes	0.20	-0.058	-0.45	-4.9
	No	-0.27	-0.47	-0.21	0.18
	None	-2.6	-2.8	-2.5	-2.1
Vacuum	None	-2.3	-2.5	-2.2	-1.9
					2.6

Experimental decane $\Delta G_{\text{solv}} = 3.16 \text{ kcal/mol}$

Using Parameterized Molecules to Test FEP Methods

Testing FEP Method with Parameterized CHARMM Test Set

Compound	ΔG_{solv} (kcal/mol)		Compound	ΔG_{solv} (kcal/mol)	
	Calc.	Exp. ⁵		Calc.	Exp. ⁵
acetic acid	-12	-6.69	methanol	-4.0	-5.10
benzene	-0.82	-0.86	methylamine	-2.7	-4.55
butane	1.8	2.07	N-methylacetamide	-12	-10.00
ethanol	-6.0	-5.00	pentane	1.1	2.32
ethane	2.2	1.83	phenol	-7.8	-6.61
ethanethiol	-0.98	-1.10	propane	2.2	1.96
methanethiol	-0.35	-1.20	protene	0.80	1.32

Average Deviation from experiment: 1.2 kcal/mol

Exp. Values: Rizzo, R.C. et al. **2006**. *J. Chem. Theor. Comp.* 2: 128-139.

Future Work

- Continue to refine TCA CHARMM parameters in a more systematic manner
 - Use new parameters to perform calculations that can be compared to experimental results (e.g., pKa prediction)
- Investigate alternative methods for implementation of FEP calculations into NAMD, including those that can separate electrostatic decoupling from van der Waals
- Use proper TCA parameters to perform FEP calculations that simulate the mutation of an inhibitor-binding residue and compare these results with mutagenesis studies. This data can be used to refine the computational model of TCA binding to both LeuT_{Aa} and mammalian sodium symporters (e.g., DAT).

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